

# 1 Microphysical controls on the isotopic composition of 2 wintertime orographic precipitation

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## Key Points

- 3 • Distinct isotopic signatures of microphysical processes can be determined.
- 4 • Temperature and mountain height control precipitation isotopic composition more strongly  
5 than CDNC.
- 6 • Microphysical changes with CDNC, temperature, and mountain height establish isotopic  
7 composition.

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8 **Abstract.** The sensitivity of mixed-phase orographic clouds, precipita-  
9 tion and their isotopic content to changes in dynamics, thermodynamics and  
10 microphysics is explored in idealized two-dimensional flow over a mountain  
11 barrier. These simulations use the Weather Research and Forecasting (WRF)  
12 Model with stable water isotopologues (HDO and H<sub>2</sub><sup>18</sup>O), which have been  
13 integrated into the Thompson microphysics scheme within WRF as part of  
14 the present project. In order to understand how the isotopic composition of

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15 precipitation ( $\delta^{18}\text{O}_{precip}$ ) is fixed, the mountain height, temperature, and the  
16 prescribed cloud droplet number concentration (CDNC) have been varied  
17 in a series of simulations. For the given range of values explored in this work,  
18 changes in mountain height and temperature induce stronger responses in  
19 domain-averaged  $\delta^{18}\text{O}_{precip}$  than do changes in CDNC by a factor of approx-  
20 imately 10. The strongest response to changing CDNC leads to local vari-  
21 ations of  $\delta^{18}\text{O}_{precip}$  of about 3‰, though those occur in regions of weak pre-  
22 cipitation ( $<0.1 \text{ mm hr}^{-1}$ ). Changes in  $\delta^{18}\text{O}_{precip}$  can be understood through  
23 the microphysical pathways by which precipitable hydrometeors are formed  
24 and by the isotopic signature associated with each pathway. The decrease  
25 in  $\delta^{18}\text{O}_{precip}$  with increasing mountain height is not just a function of decreas-  
26 ing temperature, but also reflects the changing contributions and distinct iso-  
27 topic signatures of riming of cloud liquid and vapor deposition onto snow,  
28 the leading sources of precipitation in these simulations. The changes in  $\delta^{18}\text{O}_{precip}$   
29 with mountain height, temperature and CDNC are governed in part by the  
30 microphysical pathways through which precipitating hydrometeors are formed  
31 and grow.

## 1. Introduction

32 Precipitation that forms due to interaction with mountain barriers, or orographic pre-  
33 cipitation, is an important contributor to surface water resources. In particular, runoff  
34 from rainfall and melting of the mountain snowpack feed into river basins that provide  
35 water to a number of heavily populated regions. As the amount and location of precip-  
36 itation on the mountain barrier will determine the volume of runoff and the watershed  
37 into which it flows, understanding all of the factors that influence and control orographic  
38 precipitation is essential for current and future forecasts of this necessary resource.

39 There have been extensive regional studies regarding the formation and behavior of  
40 orographic precipitation [e.g. *Hobbs*, 1975; *Smith et al.*, 2005; *Smith and Evans*, 2007;  
41 *Zubler et al.*, 2011]. The total precipitation and its spatial distribution have been found to  
42 be dependent upon several variables, including the orientation and geometry of the terrain,  
43 atmospheric stability, orographic flow dynamics and cloud microphysics [e.g. *Colle*, 2004;  
44 *Galewsky*, 2008; *Muhlbauer and Lohmann*, 2008]. In terms of cloud microphysics, the  
45 different pathways through which precipitating hydrometeors grow can be more or less  
46 efficient and thus greatly influence the amount of precipitation. For example, in mixed-  
47 phase orographic clouds, the growth and fallout of snow and graupel may be enhanced by  
48 the “seeder-feeder” mechanism [*Reinking et al.*, 2000], wherein ice crystals grow by vapor  
49 deposition in an ice cloud aloft before sedimenting to lower levels in the cloud where  
50 the ice continues to grow by collecting cloud droplets (riming). This enhanced low-level  
51 riming increases the fallspeed of snow and also the overall precipitation efficiency of the  
52 cloud [*Mitchell et al.*, 1990; *Borys et al.*, 2003], thereby augmenting precipitation on the

53 windward side of the mountain at the expense of the transport of hydrometeors to the  
54 leeward slope (and the resulting precipitation there, which is known as “spillover”).

55 The stable isotopologues of water ( $\text{H}_2^{16}\text{O}$ ,  $\text{HDO}$ ,  $\text{H}_2^{18}\text{O}$ ) have been used in precipita-  
56 tion analysis dating back to the initial work of *Dansgaard* [1952]. In the mid-latitudes,  
57 where westerlies impinging on north-south oriented mountain ranges form the motiva-  
58 tion for our idealized simulations, the isotopic composition of precipitation is primarily  
59 temperature-dependent [e.g. *Dansgaard*, 1964; *Noone and Simmonds*, 2002; *Jouzel*, 2003;  
60 *Lee et al.*, 2007], such that the ratio of the heavy (e.g.,  $\text{H}_2^{18}\text{O}$ ) to light ( $\text{H}_2^{16}\text{O}$ ) isotopes  
61 correlates positively with temperature. In mountainous regions, the relationship between  
62 the isotopic composition of precipitation and temperature is additionally linked with alti-  
63 tude [*Dansgaard*, 1964]. Air cools as it rises along the upslope on the windward side of a  
64 mountain, and the progressive removal of precipitation produces a gradient in the isotopic  
65 composition with altitude. This leads to precipitation enriched in heavy isotopes forming  
66 at lower altitudes, and more depleted precipitation (i.e., with lower isotopic ratios) at  
67 higher altitudes, as well as on the downslope in the lee of the mountain peak [*Smith et*  
68 *al.*, 2005]. This isotopic gradient was connected to the fractional removal of water by a  
69 mountain barrier and its drying ratio by *Smith et al.* [2005]. The relationship between  
70 isotopic composition and altitude has also been used to relate paleoclimate proxies for the  
71 isotopic composition of precipitation to past mountain elevation [*Poage and Chamberlain*,  
72 2001; *Rowley et al.*, 2001]. However, as demonstrated by *Galewsky* [2009] and *Lechler and*  
73 *Galewsky* [2013], in different dynamical regimes, the airflow over the mountain can com-  
74 plicate the relationship between the isotopic composition of precipitation and the altitude  
75 of a mountain barrier.

76 In addition to the dynamical influences on orographic precipitation and its isotopic  
77 content, microphysical processes can also modify the isotopic signature of precipitation.  
78 *Coplen et al.* [2015] connected variations in the isotopic content of precipitation in land-  
79 falling extratropical cyclones with changes in the storm structure and different pathways  
80 of precipitation formation. Observations of snowfall from the Sierra Nevada [*Demoz et al.*,  
81 1991] and snowfall and cloud liquid in Colorado [*Lowenthal et al.*, 2011] suggested that the  
82 isotopic composition of snowfall is influenced by the degree of riming. By sampling both  
83 the isotopic and chemical composition of both snowfall and cloud droplets at a mountain-  
84 top site in Colorado, *Lowenthal et al.* [2011] related the degree of riming of snowfall to the  
85 chemical composition of the snow and concurrently sampled cloud droplets. They found  
86 that snow mass formed mainly through riming was more enriched and had an isotopic  
87 signature that was similar to the cloud droplets. This relationship was then employed to  
88 make predictions about the altitude at which snow formed through vapor deposition.

89 The relative role of riming in mixed-phase orographic precipitation can be reduced by  
90 decreasing temperature, through the glaciation of liquid clouds, and by increasing aerosol  
91 concentrations, which tend to lead to more numerous and smaller cloud droplets that  
92 are less likely to be collected by falling snow [*Pruppacher and Klett*, 1997; *Wang and*  
93 *Ji*, 2000]. Increased aerosol concentrations can also suppress or delay the formation of  
94 precipitation in liquid-only clouds by reducing the efficiency of collision and coalescence  
95 processes [*Albrecht*, 1989; *Ramanathan et al.*, 2001]. While aerosols have the potential  
96 to impact individual microphysical processes that contribute to precipitation, their influ-  
97 ence on the amount and distribution of orographic precipitation has not been definitely  
98 established and appears to depend strongly on the environmental conditions of the region

99 being considered [*Borys et al.*, 2000, 2003; *Khain and Pokrovsky*, 2004; *Lynn et al.*, 2007;  
100 *Muhlbauer and Lohmann*, 2008; *Saleeby et al.*, 2013]. In a study of warm (liquid-only)  
101 orographic clouds and precipitation, *Miltenberger et al.* [2015] suggested that interactions  
102 between dynamical and microphysical processes can lead to regimes where the precipita-  
103 tion and the precipitation efficiency are insensitive to changes in cloud droplet number  
104 concentration (CDNC), which is used as a proxy for aerosol concentrations.

105 To explore how such changes in microphysical processes influence mixed-phase oro-  
106 graphic precipitation and its isotopic content, we perform a number of idealized two-  
107 dimensional simulations in which the mountain height, temperature and CDNC are var-  
108 ied. Particular attention is paid to changes in the microphysical processes that contribute  
109 to the growth of precipitating hydrometeors and how those processes and their isotopic  
110 signatures control the amount, distribution and isotopic content of precipitation in these  
111 experiments. By tracking the isotopic ratio associated with precipitation growth processes  
112 within these simulations, we determine if each microphysical process has a distinct isotopic  
113 signature and how each process contributes to the overall isotopic signal of precipitation  
114 in different regimes.

115 Although the results presented in this paper are based on idealized simulations, they  
116 represent a step towards constructing an isotope-enabled regional modeling capability for  
117 WRF. Previous work with isotope-enabled global models [e.g. *Noone and Simmonds*, 2002;  
118 *Vuille et al.*, 2003; *Lee et al.*, 2007; *Field*, 2010] has advanced our knowledge of how large-  
119 scale processes affect isotopic composition. However, the limitations of global climate  
120 models (GCMs) in representing topography and cloud-scale processes leaves room for  
121 higher-resolution regional models that more faithfully represent such fine-scale phenomena

122 [*Prein et al.*, 2015]. The work of *Pfahl et al.* [2012] provides an example of how a fine-scale  
123 regional model can improve the representation of isotopic signals over that of an isotope-  
124 enabled GCM. By using a regional model with horizontal grid spacing of  $O(1\text{ km})$ , this  
125 work aims to understand orographic precipitation resolving scales much finer than those  
126 represented in a typical isotope-enabled GCM whose horizontal grid spacings vary from  
127 approximately 200 to 400 km [*Conroy et al.*, 2013].

## 2. Model and Experiments

### 2.1. Model Setup

128 To conduct the orographic precipitation experiments, we use the WRF model version  
129 3.5.1 [*Skamarock and Klemp*, 2008] provided by the Mesoscale and Microscale Meteorol-  
130 ogy Division of the National Center for Atmospheric Research. The model is configured  
131 to perform simulations of idealized 2D flow over a hill. The domain consists of 300 grid  
132 points with 2 km spacing in the horizontal direction and 105 vertical levels whose spacing  
133 varies from 25-200 m in the lower 5 km and is uniform above 5 km. The duration of each  
134 simulation is twelve hours. The Thompson microphysics scheme [*Thompson et al.*, 2008]  
135 was chosen for this work because a study of wintertime precipitation in a mountainous re-  
136 gion of the western United States found that, along with one other scheme, the Thompson  
137 microphysics scheme provided the best representation of cold season snowfall [*Liu et al.*,  
138 2011]. In addition, the Thompson scheme includes a detailed treatment of the riming of  
139 cloud droplets by snow (as in *Saleeby and Cotton* [2008]), which has proved important for  
140 realistic simulation of the effects of pollution on riming in mixed-phase orographic clouds  
141 [*Lohmann*, 2004; *Saleeby and Cotton*, 2008; *Saleeby et al.*, 2011].



142 As noted previously, different atmospheric regimes are simulated in order to study the  
143 response of model microphysics. The Thompson microphysics scheme allows the user to  
144 specify the CDNC value, which is utilized in this work as a proxy for aerosols. The chosen  
145 CDNC values represent conditions that range from pristine to polluted. The working  
146 assumption is that for high aerosol loading, there are more cloud condensation nuclei  
147 (CCN) and thus a higher CDNC value, while a lower CDNC value indicates a scenario  
148 with few aerosols and therefore fewer CCN. More specifics about the experimental setup  
149 are given in section 2.3.

## 2.2. Isotopic implementation

150 In its default configuration, the Thompson scheme only treats microphysical transfers of  
151 the standard isotopologue of water ( $\text{H}_2^{16}\text{O}$ ) among water vapor and the different hydrom-  
152 eteors included in the scheme: cloud liquid, rain, cloud ice, snow and graupel. As part of  
153 the present project, we have extended the Thompson scheme so that the microphysical  
154 transfers of the stable isotopologues of water (HDO and  $\text{H}_2^{18}\text{O}$ ) are also included. The  
155 isotopic composition of water vapor and each hydrometeor is tracked, and the exchanges  
156 of the heavy isotopologues of water are accounted for during each microphysical process  
157 represented in the Thompson scheme. Isotopic fractionation — the unequal exchange of  
158 heavy and lighter isotopologues of water — is accounted for in processes that involve the  
159 deposition of vapor onto liquid or ice hydrometeors and those involving the evaporation of  
160 liquid phase hydrometeors (rain or cloud liquid). Other processes that involve the trans-  
161 fer of whole hydrometeors from one category to another (e.g., freezing, melting, riming),  
162 occur without fractionation. As in *Bony et al.* [2008], *Blossey et al.* [2010] and *Pfahl et*  
163 *al.* [2012], the sublimation of ice phase hydrometeors (snow, cloud ice, graupel) is also

164 assumed to occur without fractionation, so that the vapor produced by sublimation of  
165 snow, for example, has the same isotopic composition as the snow. While sublimation  
166 is expected to produce vapor from the outer shell of an ice phase hydrometeor, and this  
167 layer may not have the same isotopic composition as the particle as a whole, tracking the  
168 composition of individual layers within the crystals is judged to be too complicated and  
169 expensive to include in the present implementation. Note that the implementation, which  
170 follows *Blossey et al.* [2010], includes few approximations in its representation of isotopic  
171 exchanges beyond the assumption that the isotopic composition of each hydrometeor cat-  
172 egory in a given grid cell is uniform and that no fractionation occurs during sublimation.  
173 The model uses time steps on the order of a few seconds, so that only cloud liquid and  
174 vapor are assumed to equilibrate within a single time step. Other processes are integrated  
175 in time explicitly by the model. A more detailed description of the water isotope physics  
176 is given in Appendix A.

177 The quality of the isotopic simulation depends strongly on the representation of the  
178 standard isotopologue of water. If the microphysics scheme and the broader model do a  
179 poor job in representing the amount and distribution of precipitation of the standard iso-  
180 topologue of water, this will be reflected in the isotopic composition as well. Encouraged  
181 by the performance of the Thompson scheme within WRF on wintertime orographic pre-  
182 cipitation [*Liu et al.*, 2011] and by the representation of isotopic composition in tropical  
183 convection in a similar implementation of water isotopologues in *Blossey et al.* [2010], we  
184 proceed with the simulations here.

185 For the isotopic analysis, our results on  $\text{H}_2^{18}\text{O}$  are presented in delta-notation such  
186 that  $\delta^{18}\text{O} = 1000 \left( \frac{R}{R_o} - 1 \right)$ , where  $R$  is the isotopic ratio of  $\text{H}_2^{18}\text{O}$  in a specified water

187 species and  $R_o$  is the isotopic ratio of the standard. While HDO is also included in the  
188 microphysics scheme, the additional information that can be gained by considering both  
189 HDO and  $H_2^{18}O$  will be left to future work.

### 2.3. Experimental Setup

190 Several experiments are conducted that alter the initial temperature profile, mountain  
191 height (800, 1500 and 3000 m), and the CDNC (25, 100, 200, 400, and 800  $cm^{-3}$ ). Two  
192 initial temperature profiles are used here and are referenced as warm or cold based on the  
193 surface temperature of the upstream sounding ( $T_{sfc} = 7^\circ C$  and  $0^\circ C$ , respectively). Ex-  
194 periments are referenced by abbreviations (e.g., W800m), which indicate the temperature  
195 sounding (W=warm or C=cold) and mountain height settings. The setup and initial con-  
196 ditions, including the temperature profiles, are similar to those in *Muhlbauer et al.* [2010],  
197 with a mountain half-width of 20 km and a horizontal wind profile that is a constant  
198  $15\text{ m s}^{-1}$  below 10 km and linearly increases to  $40\text{ m s}^{-1}$  at the top model layer (30 km).

199 To generate the initial vapor conditions for  $H_2^{18}O$  and HDO, a Rayleigh distillation  
200 profile is generated assuming equilibrium with ocean water at  $20^\circ C$ , which represents the  
201 average temperature of the ocean surface where the initial isotopic signature of the air  
202 mass will be set. The model's initial conditions for the isotopic content of water vapor are  
203 interpolated from this Rayleigh profile based on the water vapor mass mixing ratio. As  
204 the cold sounding is drier than the warm sounding, it is also more depleted, such that the  
205  $\delta^{18}O$  of vapor at the surface is 8‰ less than that of the warm sounding. Neither liquid  
206 nor ice condensate exists initially, and therefore their isotopic compositions do not need  
207 to be initialized.

## 2.4. Model Validation

208 The model's ability to simulate orographic clouds and precipitation is on par with previ-  
209 ous studies. The results are very similar to the WRF simulations in *Muhlbauer et al.* [2010],  
210 despite the use of a different microphysical scheme. There are some small differences in  
211 the simulated orographic clouds, and our experiments produce more accumulated precip-  
212 itation. However, these deviations can be attributed to our implementation of a longer  
213 simulation time and a larger range of CDNC values in addition to the choice of microphys-  
214 ical scheme. The changes in the liquid orographic cloud are also similar to results seen  
215 by *Xiao et al.* [2014], who also used the same idealized WRF setup, but coupled with a  
216 detailed bin microphysics scheme and a warmer initial temperature profile. The evolution  
217 of cloud liquid and microphysical processes as CDNC increases in our cold experiments is  
218 similar to that of *Saleeby et al.* [2006], who used the Colorado State University - Regional  
219 Atmospheric Modeling System with a different microphysics scheme to simulate realistic  
220 wintertime orographic clouds in northern Colorado.

221 This project represents the first use of this isotope-enabled version of the Thompson  
222 scheme within WRF. As the present modeling study is idealized and the incorporation of  
223 water isotopologues into the real-case forecasting capability of WRF is not complete, we  
224 focus on the performance of the scheme within the present simulations. First, the isotopic  
225 composition of precipitation along the upslope of the mountain approximately conforms to  
226 that of a Rayleigh process and is slightly more depleted than the Rayleigh process due to  
227 dynamical effects of the mountain [*Galewsky, 2009*] and the formation of precipitation from  
228 more depleted vapor above the surface of the mountain (supplemental Fig. S1). Second,  
229 in section 3.3, closed budgets for the surface precipitation are constructed that explain

230 the precipitation itself and its isotopic composition in terms of the various microphysical  
231 processes that contribute to the formation and growth of precipitating hydrometeors.  
232 Last, the isotopic composition of water vapor and hydrometeors described in sections 3.1  
233 and 3.4 shows the expected influence of microphysical processes on isotopic composition,  
234 such as isotopic equilibration of cloud liquid and water vapor, vapor deposition onto ice  
235 and the evaporation of rain in subsaturated conditions.

236 It should also be noted that the average  $\delta^{18}\text{O}$  values of total precipitation compare well  
237 with observations in conditions similar to those used here for initial conditions. *Ander-*  
238 *son et al.* [2015] calculated the average  $\delta^{18}\text{O}$  of snowpack using the Isotopes in Rocky  
239 Mountain Snowpack (IRMS) database, and found that values ranged between  $-10\text{‰}$  and  
240  $-25\text{‰}$ , which compares well with the range of  $\delta^{18}\text{O}$  in the average precipitation for our  
241 experiments (approximately  $-10\text{‰}$  to  $-16\text{‰}$  in warm simulations and  $-19\text{‰}$  to  $-26\text{‰}$  in  
242 cold experiments). Comparable  $\delta^{18}\text{O}_{precip}$  values ( $-12\text{‰}$  to  $-24\text{‰}$ ) were measured during  
243 a 1985 March storm in Kingvale, CA, which is located upwind of the Sierra Nevada crest  
244 at an elevation of 1859 m [*Warburton et al.*, 1993]. *Warburton and DeFelice* [1986] ana-  
245 lyzed samples in the Central Sierra Nevada, and found that snow formed through vapor  
246 deposition had a  $\delta^{18}\text{O}$  signature that ranged from  $-18.4\text{‰}$  to  $-22.9\text{‰}$ , which corresponds  
247 well with our cold temperature profile experiments (see further discussion in section 3).  
248 The snow samples from the same study that indicated growth by a combination of riming  
249 and vapor deposition, were less depleted and ranged between  $-6.4\text{‰}$  and  $-16.8\text{‰}$ , which  
250 resembles results in our warm simulations (see section 3). Values similar to *Warburton*  
251 *and DeFelice* [1986] were measured in Colorado by *Lowenthal et al.* [2011] for snow that  
252 had undergone little riming. In the same study, snow that experienced more riming (as in-

253 dicated by higher concentrations of sulfate), was less depleted and ranged between  $-15.6\text{‰}$   
254 and  $-20.4\text{‰}$ .

### 3. Model Results

#### 3.1. Reference Simulation

255 To outline the general characteristics of the cloud and precipitation in these simulations,  
256 the simulation with the warmer sounding ( $T_{sfc} = 7^{\circ}\text{C}$ ), a 800 m high mountain, and a cloud  
257 droplet number concentration (CDNC) of  $200\text{ cm}^{-3}$  is chosen as the reference simulation.  
258 Fig. 1b shows the average simulated mass of cloud liquid along with the combined mass  
259 of cloud ice and snow for the reference simulation. (Figs. 1a and 1c will be discussed  
260 in section 3.2.) The figure combines cloud ice and snow together, as the setup of the  
261 Thompson scheme quickly leads to the conversion of cloud ice to snow, and as a result,  
262 produces little cloud ice [*Thompson et al.*, 2008]. Note that while a wave cloud exists  
263 aloft and downstream of the mountain in these simulations, our focus is on the cloud and  
264 precipitation over the mountain, where almost all precipitation is produced.

265 The orographic cloud in the reference simulation does not extend higher than 4 km and  
266 has a much higher mass of cloud liquid than a combined mass of snow and ice (Fig. 1b).  
267 The frozen hydrometeors occur predominately upstream of the mountain peak with only  
268 a little spillover ( $\sim 10$  km) to the downstream side. For the most part, the liquid and  
269 ice/snow regions of the cloud overlap, except on the leeward slope, where the glaciated  
270 cloud is located above the liquid one.

271 The isotopic values of vapor, cloud liquid, rain and ice/snow for the reference simu-  
272 lation are presented in Fig. 2. Cloud liquid isotopic values range from approximately  
273  $-7\text{‰}$  near the mountain surface to  $-22\text{‰}$  at cloud top (Fig. 2c). Isotopic equilibrium is

274 enforced between cloud liquid and vapor, so that decreasing  $\delta^{18}\text{O}$  of cloud liquid with  
275 height is expected given that the vapor  $\delta^{18}\text{O}$  shows the same trend (Figs. 2a and 2c). The  
276 cloud liquid that extends further leeward has roughly the same  $\delta^{18}\text{O}$  value as the cloud  
277 liquid on the corresponding windward side, so there is no obvious  $\delta^{18}\text{O}$  difference between  
278 the windward and leeward cloud liquid. As expected from rainout (i.e., the progressive  
279 removal of heavy isotopologues by precipitation across the mountain barrier [*Clark and*  
280 *Fritz, 1997; Smith et al., 2005*]), the  $\delta^{18}\text{O}$  of water vapor does show asymmetry about  
281 the mountain and is more depleted at low levels further downstream of the mountain.  
282 The cloud ice/snow  $\delta^{18}\text{O}$  values range from  $-10\text{‰}$  near cloud base to  $-35\text{‰}$  at cloud top  
283 (Fig. 2d).

284 Fig. 3a shows the profile of accumulated precipitation across the mountain for the refer-  
285 ence simulation along with a number of different CDNC concentrations. (The sensitivity  
286 to CDNC will be discussed in the following section.) The precipitation for the reference  
287 simulation ( $\text{CDNC} = 200 \text{ cm}^{-3}$ , red line) peaks over the mountain top and is nearly sym-  
288 metric, with slightly more precipitation falling downwind of the peak and a spillover ratio  
289 of 0.56 (Tab. 1). (The spillover ratio is the ratio of the accumulated leeward precipitation  
290 to total precipitation.) Most of the precipitation falls as rain at the surface, with similar,  
291 smaller amounts of snow and graupel (Tab. 1). The isotopic composition of the accumu-  
292 lated precipitation  $\delta^{18}\text{O}_{precip}$  in the reference simulation (Fig. 3b, red line) becomes more  
293 enriched as one ascends the lower slope on the upwind side of the mountain. This is also  
294 seen in the isotopic composition of rain in Fig. 2b, and is associated with a shift from rain  
295 resulting from the melting of snow that was formed aloft through vapor deposition, to rain  
296 and snow that grew through the conversion and accretion of cloud liquid. Such changes

297 in the microphysical pathways through which precipitating hydrometeors are formed and  
298 their impact on  $\delta^{18}\text{O}_{precip}$  will be discussed in greater detail in section 3.3. Following this  
299 peak in  $\delta^{18}\text{O}_{precip}$  at  $x=280$  km, the isotopic composition of precipitation falls off across  
300 the mountain as the heavier isotopes are removed preferentially through fallout. As noted  
301 in the introduction, this may be modeled approximately as a Rayleigh process [*Smith et*  
302 *al.*, 2005], though there are some complications due to dynamical response to topography  
303 [*Galewsky*, 2009] and microphysical effects. The increase in  $\delta^{18}\text{O}_{precip}$  on the downslope  
304 at  $x=315-320$  km is associated with the fractionation of evaporating rain once it passes  
305 downstream of the orographic cloud (see also Fig. 2b-c.) As shown in previous studies  
306 [e.g. *Stewart*, 1975; *Lawrence et al.*, 1998; *Bony et al.*, 2008; *Risi et al.*, 2008], evaporation  
307 in subsaturated conditions tends to enrich the rain and deplete the vapor, as the lighter  
308  $\text{H}_2\text{O}$  will more quickly move from the liquid to the surrounding vapor.

### 3.2. Sensitivity to CDNC

309 Next, the sensitivity of the reference simulation to changes in CDNC (as a proxy for  
310 aerosol variations) is shown. This is interesting both as a way to understand whether  
311 aerosol impacts on orographic precipitation [e.g., *Rosenfeld et al.*, 2008] could impact  
312 the isotopic composition as well, and as an example of how changing the microphysical  
313 processes which contribute to precipitation could impact the amount, distribution and  
314 isotopic composition of orographic precipitation.

315 Three cases with increasing values of CDNC are shown in Fig. 1, which illustrates  
316 potential changes in the orographic cloud with CDNC. As the CDNC value increases, the  
317 conversion of cloud to rain and the riming of cloud liquid by snow become less efficient,  
318 resulting in an increase in both the amount of cloud liquid and the leeward region it spans.



319 While there is already leeward spillover of cloud liquid in the  $25 \text{ cm}^{-3}$  case, cloud liquid  
320 extends an additional 15 km down the leeward side in the  $800 \text{ cm}^{-3}$  experiment, with the  
321 region of maximum mass mixing ratio (red filled contours) also reaching approximately  
322 5 km further downstream. This shift in the leeward extent of cloud liquid is mirrored in  
323 the isotopic composition of precipitation in Fig. 3b, where the increase in  $\delta^{18}\text{O}_{precip}$  due to  
324 rain evaporation occurs farther downstream as CDNC increases. On the windward slope,  
325 the location of the leading edge of the cloud does not change in all of the warm 800 m  
326 experiments. Increases in CDNC have little impact on the location of snow and cloud  
327 ice: both the horizontal and vertical extent of the glaciated cloud remain the same. The  
328 mass mixing ratio, however, does decrease very slightly (note change in contours over the  
329 mountain peak region) as the CDNC increases, which is opposite to and of much smaller  
330 magnitude than the trend found for cloud liquid. The isotopic signatures of cloud liquid  
331 and combined cloud ice/snow are similar to those of the reference simulation (Fig. 2) and  
332 are not shown.

333 As in the reference simulation, most of the precipitation in the simulations with vary-  
334 ing CDNC falls as rain (see Tab. 1) with small, similar amounts of accumulated snow  
335 and graupel. The third column in Tab. 1 indicates that the accumulated precipitation  
336 decreases as CDNC increases, and is reduced by more than half between the  $25 \text{ cm}^{-3}$   
337 and  $800 \text{ cm}^{-3}$  experiments. Fig. 3a shows that the location of the maximum precipita-  
338 tion shifts leeward as CDNC increases, which has been previously observed in wintertime  
339 orographic precipitation [*Jirak and Cotton, 2006; Saleeby et al., 2011*]. Among the dif-  
340 ferent mountain heights and temperatures considered here, the magnitude of the shift  
341 is strongest and most obvious for the W800m experiments, where there is a difference

of approximately 10 km between the precipitation peaks in the  $25 \text{ cm}^{-3}$  and  $800 \text{ cm}^{-3}$  simulations. This shift is also evident in the spillover calculations in Tab. 1.

The domain can be broken down into three smaller regions: upstream of the peak (up to 290 km into the domain), around the peak (290–310 km) and downstream of the peak (310 km onwards). As Fig. 3a indicates, most of the precipitation falls in the first and second regions. The influence of CDNC on precipitation is also most pronounced in these regions. However, the CDNC impact on the  $\delta^{18}\text{O}_{precip}$  is slightly different. Fig. 3b illustrates that the largest  $\delta^{18}\text{O}_{precip}$  difference between simulations occurs in the first region, but over the second region, variation in the isotopic signal is small ( $\leq 1\%$ ). In the third region, downstream of the mountain peak, the accumulated precipitation is relatively unchanged between simulations, but there is some separation in the  $\delta^{18}\text{O}_{precip}$  of approximately 2‰ at  $x=310$  km before the effects of rain evaporation enter further down the lee slope. The slopes of  $\delta^{18}\text{O}_{precip}$  across the peak differ, with the steepest change in  $\delta^{18}\text{O}_{precip}$  across the peak in the simulation with the largest precipitation (CDNC= $25 \text{ cm}^{-3}$ ) as one would expect due to the effect of rainout [Smith *et al.*, 2005]. This leads the  $25 \text{ cm}^{-3}$  simulation to have the largest upstream-downstream difference in  $\delta^{18}\text{O}_{precip}$  around the peak ( $x=290\text{--}310$  km).

### 3.3. Microphysical pathways

To better understand the changes in precipitation and its isotopic composition across the mountain, we consider the budget for the total mass of precipitating hydrometeors (rain, snow and graupel combined) in these simulations, integrated in time and over the whole domain or a sub-region of the domain. Since isotopic composition is unchanged by exchanges between rain, snow and graupel by freezing, melting or aggregation, we focus

364 on the sources which determine the isotopic composition of the precipitation: autocon-  
 365 version/accretion of cloud liquid or cloud ice, riming of cloud liquid and exchanges with  
 366 vapor by deposition or sublimation/evaporation. In this budget, surface precipitation,  $P$ ,  
 367 is a sink of hydrometeor mass and is balanced by various microphysical sources of rain,  
 368 snow and graupel as well as advection and storage of these hydrometeors:

$$P = Q_{LAUT} + Q_{LACC} + Q_{IAUT} + Q_{IACC} + Q_{RIM} +$$

$$Q_{DEP} + Q_{SUB} + Q_{ADV} - Q_{STOR}. \quad (1)$$

370 Here, the sources of hydrometeor mass include microphysical processes, such as autocon-  
 371 version of cloud liquid (LAUT), accretion of cloud liquid (LACC), autoconversion of cloud  
 372 ice (IAUT), accretion of cloud ice (IACC), riming of cloud liquid (RIM), vapor deposi-  
 373 tion onto ice (DEP), sublimation of ice (SUB), along with those associated with moisture  
 374 flux convergence (labeled ADV for advection) and storage (STOR). The storage term is  
 375 negative because increases in hydrometeors in the domain over time come at the expense  
 376 of surface precipitation. Each of these terms are integrated over the duration of the sim-  
 377 ulations and over the domain or a subset of the domain in the horizontal direction and  
 378 then normalized by the mountain half-width (20 km). Note that, because the Thomp-  
 379 son microphysical scheme produces little cloud ice, much of the vapor deposition onto ice  
 380 phase hydrometeors that occurs in the domain contributes directly to snow growth. Other  
 381 microphysical schemes would likely have stronger vapor deposition onto cloud ice, so that  
 382 the autoconversion/accretion of cloud ice would be relatively more important and vapor  
 383 deposition relatively less important. Also, note that the net tendency of vapor deposi-  
 384 tion (including deposition, sublimation and rain evaporation) has been averaged over the  
 385 simulation and then partitioned into regions of deposition and sublimation/evaporation  
 386 according to the sign of the mean tendency.

387 A similar budget can be written for the mass of the heavy isotopologues, and the isotopic  
388 composition of those contributions can be computed from the ratio of the contribution  
389 to heavy isotope mass, e.g.,  $\text{H}_2^{18}\text{O}$ , to that for the standard isotope,  $\text{H}_2^{16}\text{O}$ . The  $\delta^{18}\text{O}$  of  
390 hydrometeor mass generated by each process may then be computed as for precipitation  
391 itself,  $\delta^{18}\text{O}_{precip}$ .

392 In Fig. 4, the total precipitation and the contributions of the dominant microphysical  
393 processes to precipitation and its isotopic composition are shown in three regions: the peak  
394 ( $x=295\text{--}305$  km) and the regions upwind and downwind of the peak. In the following,  
395 the sources of precipitation in each region are analyzed. Note that the precipitation  
396 produced in each region may fall to the surface there or be transported downstream. In  
397 Figs. 4b–d, the contribution of each process in each region has been normalized by the  
398 total precipitation in the domain for each case. These normalized contributions can be  
399 interpreted as weights, which can be applied to the characteristic isotopic composition  
400 from each process to determine  $\delta^{18}\text{O}_{precip}$ .

401 In the upwind region (Fig. 4b), riming of cloud droplets contributes most to the growth  
402 of precipitating hydrometeors, with vapor deposition onto ice making the second largest  
403 contribution in most cases. The  $25\text{ cm}^{-3}$  simulation differs in the importance of auto-  
404 conversion and accretion of cloud liquid. Riming and autoconversion of cloud liquid both  
405 have a direct dependence on the size of cloud droplets and therefore on CDNC. Accretion  
406 of cloud liquid may also depend indirectly on CDNC if less rain is generated through  
407 autoconversion as CDNC increases. As autoconversion and accretion of cloud liquid fall  
408 off with increasing CDNC, the contribution from vapor deposition increases. The isotopic  
409 signatures of the liquid processes are more enriched than the vapor deposition by approxi-

410 mately 4-9‰ (Fig. 4f), and thus the precipitation in the lower CDNC simulations is more  
411 enriched than the higher CDNC simulations. Overall, the microphysics explain the de-  
412 crease in both the accumulated precipitation and the isotopic content of the precipitation  
413 as seen in Fig. 3.

414 Riming of cloud liquid over the mountain peak is the largest source of precipitation in  
415 the three regions and itself produces enough hydrometeor mass to account for half of the  
416 surface precipitation in all cases except CDNC=25 cm<sup>-3</sup> (Fig. 4c). Accretion of cloud  
417 liquid and vapor deposition onto ice also contribute to precipitation over the peak. Similar  
418 to the upwind region, accretion of cloud liquid decreases with increasing CDNC, though  
419 more modestly, but riming actually increases. In this region, vapor deposition onto ice  
420 is still the most depleted source term. However, the ice produced by vapor deposition is  
421 more enriched above the peak than in the upstream region. The average  $\delta^{18}\text{O}$  differences  
422 between the ice produced by riming and vapor deposition over the peak range between  
423 2-4‰ (Fig. 4g). Therefore, the variation in the source terms of precipitation with CDNC  
424 over the peak produce little change in the  $\delta^{18}\text{O}_{precip}$  formed there, in part because the  
425 isotopic composition of the sources are more similar.

426 Precipitation production on the leeward slope derives predominantly from vapor depo-  
427 sition onto ice in addition to relatively small contributions from ice autoconversion and  
428 accretion of cloud liquid (Fig. 4d). One significant difference in this downwind region  
429 compared to the other two regions, is the presence of a large sink of precipitation mass  
430 caused by sublimation and rain evaporation. This pocket of sublimation/evaporation is  
431 expected due to subsidence and thus warming of air as it flows over the mountain peak.  
432 Though all of the microphysical source terms increase with CDNC in this region, removal

433 of precipitation due to sublimation/evaporation essentially balances out the source terms,  
434 and the accumulated precipitation remains relatively constant in the different simulations  
435 (Fig. 3a). The smaller range of the  $\delta^{18}\text{O}$  in the accumulated precipitation can be at-  
436 tributed to the similarity of the source terms in each simulation, except for that due to  
437 sublimation/evaporation which becomes more depleted with increasing CDNC.

### 3.4. Sensitivity to mountain height and temperature

438 The sensitivity of domain-integrated precipitation amount and its isotopic composition  
439 to CDNC changes was also studied for a number of mountain heights (800 m, 1500 m  
440 and 3000 m) and two temperature profiles (with  $T_{sfc} = 0^\circ\text{C}$  and  $7^\circ\text{C}$ )<sup>1</sup>. To understand  
441 how precipitation and its isotopic content are related across these simulations, Fig. 5  
442 shows their relationship when integrated over the whole domain (Fig. 5a) and over the  
443 regions upstream of the peak, over the peak and downstream of the peak (Figs. 5b–d,  
444 respectively). These regions are defined as above in section 3.3. As seen in Fig. 5a,  
445 the response of total (domain-integrated) precipitation and its isotopic content to CDNC  
446 changes — where it exists — is modest in comparison to that due to mountain height  
447 and temperature. The only significant response of  $\delta^{18}\text{O}_{precip}$  to CDNC occurs for small  
448 precipitation amounts (<5 mm) in the upwind region of W800m (Fig. 5b). Otherwise,  
449 the change in isotopic content due to temperature exceeds that due to CDNC by a factor  
450 of approximately 10 for the ranges of temperature and CDNC explored here. The weaker  
451 sensitivity of precipitation to CDNC changes with increasing precipitation is reminiscent  
452 of the work of *Muhlbauer et al.* [2010] for mixed-phase clouds and *Miltenberger et al.*  
453 [2015] for warm clouds. The possibility remains that a model setup that yields weaker  
454 precipitation might show a stronger sensitivity of precipitation to CDNC changes, as in

455 *Miltenberger et al.* [2015]. However, the change in  $\delta^{18}\text{O}_{precip}$  due to CDNC is unlikely to  
456 increase far beyond the range seen in the reference case (W800m).

457 Since the response to CDNC is weak in many cases, the present section focuses on a  
458 single CDNC value ( $200\text{ cm}^{-3}$ ) across the range of mountain heights and temperatures  
459 to understand the responses to mountain height and temperature seen in Fig. 5. The  
460 changing configuration of the orographic cloud with mountain height and temperature is  
461 shown in Fig. 6. The orographic cloud produced in the C800m experiment is quite similar  
462 to that of the reference simulation in terms of vertical extent (Fig. 6d). However, the  
463 extent of the cloud and snow on the lee slope changes with cloud liquid ending closer to  
464 the peak and the cloud ice/snow reaching farther down the slope. For the higher mountain  
465 heights (Figs. 6b-c, e-f), the liquid cloud is shallower in the colder simulations while the  
466 snow has a similar vertical extent. These higher mountain heights also produce more  
467 ice/snow than the reference simulation, and in the cold temperature experiments, there is  
468 more cloud ice/snow than liquid. Note that the 3000 m mountain wave response depends  
469 on temperature, with the isotherms downstream of the mountain suggesting a stronger  
470 downslope flow in the colder simulation.

471 Fig. 7 shows the isotopic composition of water vapor, cloud liquid and combined cloud  
472 ice/snow for the three mountain heights with the colder temperature profile ( $T_{sfc} = 0^\circ\text{C}$ ).  
473 Unlike in the reference simulation, these simulations have little rain, and its isotopic  
474 composition is not shown. As noted in section 2.3, the water vapor at the surface upwind  
475 of the mountain (Figs. 7a-c) is 8‰ more depleted than that of the reference simulation  
476 (Fig. 2a). The water vapor isotopic composition becomes increasingly asymmetric for  
477 the higher mountains due to rainout [*Smith et al.*, 2005], and the thin layer of downslope

478 flow is visible in the water vapor isotopic composition for the 3000 m mountain (Fig. 7c).  
479 The cloud liquid is almost entirely confined to the upstream side of the mountain, and its  
480 isotopic content (Figs. 7d–f) is tied to the water vapor through the assumption of vapor-  
481 liquid isotopic equilibrium. The combined cloud ice/snow (Figs. 7g–i) is more depleted  
482 than cloud liquid at the same altitude, and this difference increases with mountain height.  
483 On the lee side of the mountain, the snow reaches to the base of the mountain in each case  
484 and becomes more depleted with mountain height, as the snow has formed from vapor  
485 that either originates at higher altitudes or has been depleted through precipitation.

486 As suggested by the sensitivity of total precipitation shown in Fig. 5, the distribution  
487 of precipitation and its isotopic content across the mountain changes much more substan-  
488 tially with mountain height and temperature than with CDNC (Fig. 8). The precipitation  
489 amount increases and shifts upstream with increasing mountain height, and  $\delta^{18}\text{O}_{precip}$  on  
490 the lee slope becomes more depleted with mountain height in agreement with the snow  
491 isotopic composition shown in Figs. 7g–i. The lee slope difference in  $\delta^{18}\text{O}_{precip}$  between  
492 the 800 m and 3000 m mountain heights at  $x=315$  km reaches 11‰ and 13‰ in the  
493 warm and cold simulations, respectively. Similar differences are seen in the precipitation  
494 integrated in the lee of the peak in Fig. 5d. The stronger dependence of  $\delta^{18}\text{O}_{precip}$  on  
495 mountain height in the cold simulations mirrors that seen in total precipitation and its  
496 isotopic content in Fig. 5 and suggests that the isotopic lapse rate, the change in  $\delta^{18}\text{O}_{precip}$   
497 with altitude, itself depends on temperature.

498 As in section 3.3, the relative contributions of different microphysical pathways to the  
499 formation of precipitating hydrometeors are shown in Fig. 9b to understand better the  
500 influence of mountain height and temperature on isotopic composition, which was seen



501 in Fig. 5. Most of the precipitation occurs windward of the peak in the sensitivity  
502 simulations, and thus the source terms plotted in Fig. 9b are similar to the breakdown of  
503 upwind precipitation. A supplemental Fig. S2 shows the full breakdown of precipitation  
504 sources by region as in Fig. 4.

505 The  $\delta^{18}\text{O}_{precip}$  is the most enriched in the reference simulation (W800m) compared to  
506 all other simulations (Fig. 9c), and this is also the case where the sources of riming and  
507 accretion are largest and vapor deposition smallest. The contribution to precipitation from  
508 riming decreases with increasing mountain height and decreasing temperature (Fig. 9b),  
509 while the contribution of vapor deposition increases. Note that these contributions are  
510 normalized by total precipitation, which itself increases with mountain height. While  
511 there is considerable variation in the isotopic composition of the precipitation sources  
512 with mountain height and temperature, this variation is systematic in the most important  
513 contributors to precipitation: riming, vapor deposition and sublimation. As the mountain  
514 height increases or the temperature falls, these processes form precipitating hydrometeors  
515 from more depleted water vapor in the drier air found at colder temperatures and/or  
516 further aloft. Despite the variation with mountain height and temperature seen in Fig. 9d,  
517 a clear separation exists between the isotopic compositions contributed by riming and  
518 vapor deposition to precipitation, and the shift towards the formation of snow by vapor  
519 deposition at colder temperatures and higher mountains is reflected in the more depleted  
520 isotopic compositions in those experiments.

521 It is evident, particularly in the cold temperature experiments, that precipitation source  
522 significantly influences the  $\delta^{18}\text{O}_{precip}$  signal, and that the decreasing  $\delta^{18}\text{O}_{precip}$  signal with  
523 increasing mountain height is not a simple reflection of temperature. The solid line

524 in Fig. 5a–b represents the regression of the warm temperature experiments’ domain-  
525 integrated precipitation and  $\delta^{18}\text{O}_{precip}$ . The dashed line in Figs. 5a–b is the same as the  
526 solid line, but shifted down by 8‰, which represents the surface vapor  $\delta^{18}\text{O}$  difference  
527 between the warm and cold temperature profiles (see Figs. 2a and 7a). The cold 800 m  
528 simulations fall on this dashed line in Fig. 5, but as the mountain height increases, the  
529  $\delta^{18}\text{O}_{precip}$  values of the cold temperature simulations fall well below this line, implying that  
530 precipitation is more depleted than what is expected from the 8‰ offset in the upwind  
531 sounding in the cold 1500 m and 3000 m experiments. As noted above, this suggests that  
532 the dependence of  $\delta^{18}\text{O}_{precip}$  on altitude is itself a function of temperature. This can be  
533 explained by the combination of three effects. First, the changing sources of precipitation  
534 also contribute with a shift from riming to vapor deposition with decreasing tempera-  
535 ture. For the C3000m case, the domain-averaged  $\delta^{18}\text{O}_{precip}$  is close to the  $\delta^{18}\text{O}$  signatures  
536 of vapor deposition itself. Second, the nonlinearity in the relationship between isotopic  
537 composition and height plays a role here, as the gap between  $\delta^{18}\text{O}$  for the warm and cold  
538 simulations increases with height due to the curvature of the Rayleigh curve (Fig. S1).  
539 Last, the changing structure of the mountain wave with mountain height and temperature  
540 may also impact the distribution of precipitation and also its isotopic composition.

#### 4. Discussion and Conclusions

541 Orographic precipitation is an important water resource, and in this work we have at-  
542 tempted to provide new perspective on how different atmospheric regimes may influence  
543 the formation of precipitation. The isotopic composition of orographic precipitation also  
544 provides additional information about the sources of water vapor and the microphysical  
545 processes that produce this precipitation. In the present study, the microphysical controls

546 on the isotopic composition of wintertime orographic precipitation have been explored in  
547 idealized simulations of flow over a two-dimensional mountain using an isotope-enabled  
548 version of WRF. One reference simulation was performed along with sensitivity experi-  
549 ments that varied CDNC, temperature and mountain height to study the responses in the  
550 microphysical processes, their respective isotopic composition and the  $\delta^{18}\text{O}_{precip}$ .

551 One of the main goals of this work has been to study the isotopic signatures of pre-  
552 cipitation and cloud microphysical processes and determine if there is a distinct isotopic  
553 signal associated with those processes. With an idealized setup using different mountain  
554 heights, warm and cold temperature profiles, and increasing CDNC, our simulations show  
555 that there is a distinct difference in the  $\delta^{18}\text{O}$  signatures of microphysical processes. The  
556  $\delta^{18}\text{O}_{precip}$  reflects the relative contributions from each of the sources, and thus hydrome-  
557 teors that form from isotopically lighter sources lead to more depleted precipitation. In  
558 all of the simulations, precipitation grows mainly by riming of cloud liquid, vapor de-  
559 position onto ice, or a combination of the two processes. The  $\delta^{18}\text{O}$  difference between  
560 riming and vapor deposition ranges between 3-8‰ in all simulations and is independent  
561 of the environmental temperature. The distinct isotopic signals of the two sources persist  
562 despite wide variation in the isotopic composition of these sources with mountain height  
563 and temperature. This difference is related mainly to the altitude of the growth processes  
564 within the cloud, as vapor deposition occurs both near the surface and in air with more  
565 depleted water vapor aloft, and riming predominately happens near the mountain surface.

566 The sensitivity of  $\delta^{18}\text{O}_{precip}$  to mountain height and temperature reflects, in part, the  
567 changing sources of precipitating hydrometeors. The dominant source of precipitation  
568 shifts from riming for smaller mountains and the warmer temperature profile to vapor

569 deposition for higher mountains and colder temperatures. The more depleted isotopic  
570 composition of the precipitating hydrometeors generated by vapor deposition contributes  
571 to the decrease of  $\delta^{18}\text{O}_{precip}$  with mountain height and temperature. It is notable that the  
572 relationship between precipitation amount and  $\delta^{18}\text{O}_{precip}$  driven by increasing mountain  
573 height differs with temperature, and that the difference in  $\delta^{18}\text{O}_{precip}$  between the warm and  
574 cold simulations increases with mountain height. This suggests that the isotopic lapse rate  
575 is itself a function of temperature, and that this temperature dependence partly results  
576 in a shift in the microphysical pathways through which precipitating hydrometeors grow.  
577 Additional factors that could also contribute to the temperature dependence of the isotopic  
578 lapse rate include the nonlinearity of the Rayleigh curve and changes in the patterns of  
579 airflow over the mountain.

580 We have attempted to illuminate how  $\delta^{18}\text{O}_{precip}$  depends on the processes responsible  
581 for the growth of precipitating hydrometeors. The decrease in  $\delta^{18}\text{O}_{precip}$  with increasing  
582 mountain height and colder temperature profiles is largely driven by the formation of hy-  
583 drometeors from more depleted water vapor in the drier air further aloft or at colder tem-  
584 peratures. However, the pathways through which precipitating hydrometeors are formed  
585 also plays a role, as the more enriched precipitating hydrometers produced by riming  
586 contribute less to surface precipitation and the more depleted hydrometeors produced  
587 by vapor deposition onto ice contribute more. The weaker dependence of  $\delta^{18}\text{O}_{precip}$  on  
588 CDNC, where it exists, can be explained in a similar manner. While the weak dependence  
589 on CDNC suggests a similarly weak dependence on aerosol concentrations, the domain-  
590 integrated signal in the strongest case is roughly equivalent to a  $1^\circ\text{C}$  shift in temperature  
591 along the Rayleigh curve. Such a change might be visible in paleoclimate records of pre-

592 cipation if there were systematic shifts in aerosol concentrations on longer timescales.  
593 However, if the majority of precipitation was produced in colder conditions, the sensitiv-  
594 ity to CDNC might not be visible, as is the case in the more strongly precipitating cases  
595 here. While not considered here, mixed-phase orographic precipitation does respond to  
596 changes in ice nuclei concentrations [e.g., *Fan et al.*, 2014], and the associated shifts in  
597 microphysical processes could also impact the isotopic composition of precipitation.

598 The results have potential implications for research and field campaigns looking to study  
599 the influence of different atmospheric regimes on orographic precipitation, such as IFRACS  
600 (a 2014 campaign led by Doug Lowenthal, Gannet Haller and colleagues at the Desert  
601 Research Institute: [https://www.eol.ucar.edu/field\\_projects/ifracs](https://www.eol.ucar.edu/field_projects/ifracs) ), ISPA-III  
602 [*Ward and Cotton*, 2011] and StormVEx [*Mace et al.*, 2010]. As liquid processes are most  
603 responsive to CDNC, locations where precipitation primarily forms through accretion of  
604 cloud liquid and/or riming are likely to experience decreased accumulation, a shift in the  
605 location of, and a decrease in the  $\delta^{18}\text{O}_{precip}$ . However, the sensitivity depends on both the  
606 mountain height and the region above the mountain surface in which precipitation forms  
607 and grows. The model could be beneficial to those planning observational campaigns in  
608 terms of choosing locations to collect samples. For example, those interested in studying  
609 the influence of aerosols on snowfall could identify the regions where precipitation is likely  
610 to be most sensitive or least sensitive to aerosol loading.

611 In this idealized modeling study, we were able to distinguish isotopic signatures of  
612 the microphysical growth processes. As the climatology will vary between locations or  
613 even seasonally at one location, the model can be used to identify the isotopic signatures  
614 of microphysical processes in specific locations, which would help to determine growth

615 pathways of measured precipitation. The next steps are to use the isotope-enabled mi-  
616 crophysics scheme in a realistic setting to study snowfall events at Storm Peak Lab in  
617 Colorado observed during the Isotopic Fractionation in Snow (IFRACS) campaign. In  
618 this future work, we hope to expand upon our current research by studying the isotopic  
619 signatures of the microphysical growth processes that produce the observed precipitation.

## Appendix A: Incorporating Isotopologues Into the Microphysics

Stable water isotopologues were added to the Thompson microphysics scheme in the WRF model by duplicating all microphysical processes (e.g., freezing, melting, vapor deposition, evaporation) with additional process rates for the water isotopologues following *Blossey et al.* [2010, App. B]. Except for the sublimation of ice, which is assumed to be non-fractionating, all exchanges between vapor and condensate involve fractionation. The fractionation/equilibration of water isotopologues from rain is included, along with the fractionation of water vapor as it is deposited onto ice phase hydrometeors. Water vapor and cloud liquid are assumed to be in isotopic equilibrium. While a detailed description of the isotopic treatment including all of these processes can be found in appendix B of *Blossey et al.* [2010], we give a brief summary below that emphasizes those processes that play important roles in the cold and mixed-phase clouds central to this study.

For most processes, especially those in which whole hydrometeors are moved from one microphysical category to another (e.g., freezing of cloud droplets to form ice), the heavy isotopologues of water are transferred in proportion to their concentration in the source hydrometeor. For example, the freezing of cloud liquid droplets (wfz) to form cloud ice transfers heavy isotopologues to cloud ice as follows:

$$\left. \frac{dr'_i}{dt} \right|_{wfz} = \left. \frac{dr_i}{dt} \right|_{wfz} \mathcal{R}_c \quad (\text{A1})$$

where  $\mathcal{R}_c = r'_c/r_c$  is the isotopic ratio of cloud liquid, and  $r_c$  and  $r_i$  are the mass mixing ratios of cloud liquid and cloud ice, respectively. The mass mixing ratios of heavy isotopologues are denoted with primes, e.g.  $r'_c$ . Further, it is assumed that the isotopic ratio is uniform in each hydrometeor category, so that large and small raindrops have the same isotopic composition, for example. The latter assumption will not hold exactly in reality

641 and is a source of error; the computation complexity of allowing such variation in a bulk  
 642 scheme could be considerable.

643 For the few microphysical processes that result in fractionation (the unequal transfer  
 644 of heavy and light isotopologues between phases), those processes are represented as  
 645 described in *Blossey et al.* [2010, App. B]. In general, the lower vibrational energy of  
 646 the heavier isotopologues of water cause them to prefer the condensed phases (liquid, ice)  
 647 to the vapor phase, so that their concentrations in vapor are smaller than in the condensed  
 648 phases. When comparing concentrations of isotopologues, the words “heavier” or “more  
 649 enriched” are used to describe concentrations of heavy isotopologues that are higher, while  
 650 “lighter” or “more depleted” are used for smaller concentrations of heavy isotopologues.  
 651 We supply here a summary of how these processes might affect the isotopic composition  
 652 of water in mixed-phase clouds.

653 The efficient exchange between small liquid water droplets in clouds and the surrounding  
 654 water vapor leads many microphysical schemes (including Thompson) to assume that in-  
 655 cloud water vapor mixing ratios are equal to the saturation mixing ratio when cloud liquid  
 656 is present. The complementary condition for heavy isotopologues is that the isotopic ratios  
 657 of cloud liquid and water vapor are in isotopic equilibrium:

$$\mathcal{R}_c = \alpha_l \mathcal{R}_v \quad (\text{A2})$$

658 where  $\mathcal{R}_v$  is the isotopic ratio of water vapor and  $\alpha_l$  is the equilibrium fractionation  
 659 coefficient for liquid [*Majoube*, 1971]. As the equilibration time for isotopic composition  
 660 of small liquid water droplets is on the order of a few seconds [*Ciais and Jouzel*, 1994],  
 661 this is in general a good assumption and is included in our implementation.



662 Given the potentially large supersaturation with respect to ice, we need to consider the  
 663 non-equilibrium processes driven by gradients of water vapor between the environment  
 664 and ice particles which leads to vapor deposition onto those particles. The relatively  
 665 smaller diffusivities of the heavy isotopologues modifies the transfer of water to the particle  
 666 surface, so that the deposition of heavy isotopologues may be written as

$$\frac{dr'_i}{dt_{dep}} = \alpha_s \alpha_k \mathcal{R}_v \frac{dr_i}{dt_{dep}} \quad (\text{A3})$$

667 [*Ciais and Jouzel, 1994*], where  $\alpha_s$  is the equilibrium fractionation coefficient for ice [*Ma-*  
 668 *joube, 1970; Merlivat and Nief, 1967*] and the kinetic fractionation coefficient,  $\alpha_k$ , repre-  
 669 sents the effects of the relative diffusion of the heavy and light isotopologues [*Jouzel and*  
 670 *Merlivat, 1984*].

671 The effects of these two processes on the isotopic composition of liquid and ice in  
 672 mixed-phase clouds is depicted in Fig. A1. Here, the variation in the saturation ratios  
 673 with respect to liquid and ice is depicted in the left panel as a function of temperature.  
 674 In keeping with the assumption in the microphysical scheme, the saturation ratio with  
 675 respect to liquid is one, while the ice saturation ratio grows with decreasing temperature.  
 676 The isotopic content of the vapor, cloud liquid and ice formed through vapor deposition  
 677 is shown in the right panel. Equations A2 and A3 have been used to compute isotopic  
 678 composition, except for cloud liquid water, whose value is fixed to the relationship observed  
 679 in mixed-phase orographic clouds by *Lowenthal et al. [2011]*,

$$\delta^{18}\text{O}_c = 0.9T - 10.12 \quad (\text{A4})$$

680 where  $T$  is temperature in degrees Celsius. While ice formed through vapor deposition  
 681 is more enriched than cloud liquid at the same temperature, close to 0°C, increasing the

682 supersaturation with respect to ice and decreasing temperature causes a stronger kinetic  
683 effect during deposition onto cloud ice. Note that vapor deposition onto ice in a liquid  
684 class is most efficient at colder temperatures, peaking close to  $-15^{\circ}\text{C}$  [Rogers and Yau,  
685 1989, p. 161]. As a result, the typical isotopic composition of ice formed through vapor  
686 deposition is often more depleted than that of cloud liquid closer to  $0^{\circ}\text{C}$ . Note that these  
687 relationships will not hold once the liquid water is removed and the cloud is fully glaciated.

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696 should contact co-author Peter Blossey (pblossey@uw.edu).

## Notes

1. Each simulation is labeled according to its temperature and mountain height. For example, W800m denotes the reference case with “W” denoting the warmer sounding with  $T_{sfc} = 7^{\circ}\text{C}$  and, in other runs, “C” the colder sounding with  $T_{sfc} = 0^{\circ}\text{C}$ .  
697

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**Figure 1.** Contoured temperature (black) and mixing ratios of cloud liquid water (shaded) and combined cloud ice/snow (contoured) for the 800 m warm temperature simulations, averaged over the last four hours of each simulation. Cloud droplet number concentrations of (a)  $25 \text{ cm}^{-3}$ , (b)  $200 \text{ cm}^{-3}$ , and (c)  $800 \text{ cm}^{-3}$  are shown to illustrate sensitivity to CDNC. Units are  $\text{kg kg}^{-1}$  for hydrometeor mixing ratios and K for temperature.

**Figure 2.** Average over the last four hours of  $\delta^{18}\text{O}$  of (a) vapor, (b) rain, (c) cloud liquid and (d) ice/snow for reference simulation with a CDNC of  $200 \text{ cm}^{-3}$ .

**Figure 3.** (a) Precipitation accumulated over the 12 hours of the simulation and (b) the associated  $\delta^{18}\text{O}_{precip}$  for the reference simulation and its sensitivity to CDNC changes. The mountain peak is located at 300 km.

**Figure 4.** Breakdown of (a) accumulated precipitation in the reference simulation over the whole domain and the regions upwind of the peak, over the peak and downwind of the peak, as well as (e) the respective average  $\delta^{18}\text{O}$  values. The contributions from precipitation sources normalized by the total, domain-integrated precipitation in the three subregions are shown in (b–d) and the corresponding isotopic signatures in (e–g). The accumulated precipitation and its sources are integrated over the domain and normalized by the mountain half-width (20 km). Sources are: autoconversion of cloud liquid (laut), accretion of cloud liquid by rain (lacc), autoconversion of cloud ice (iaut), riming of cloud liquid (rim), vapor deposition onto ice (dep), and sublimation of ice/evaporation of rain (sub).

**Figure 5.** Scatter plot of area-integrated precipitation vs  $\delta^{18}\text{O}_{precip}$  for all mountain heights, temperatures and CDNC values. These quantities are presented both for the whole domain (a) and for regions upstream of the peak (b), over the peak (c) and to the lee of the peak (d). For each experiment, the CDNC value is indicated by the color and size of the symbol, while the mountain height and temperature are shown by the shape of the symbol. The grey solid line in (a) and (b) is the regression of the domain averages in the warm temperature experiments and the dashed line is the solid line shifted down by 8‰. The regression is not shown in (c) and (d) because the isotopic composition over and downwind of the peak depends on the precipitation amount upstream.

**Figure 6.** As in figure 1, but for the  $200\text{ cm}^{-3}$  simulations from the (a) 800 m, (b) 1500 m, (c) 3000 m warm temperature experiments and (d) 800 m, (e) 1500 m, and (f) 3000 m cold temperature experiments.

**Figure 7.** Based on averages over the last four hours of each simulation,  $\delta^{18}\text{O}$  of (top) vapor, (middle) cloud liquid and (bottom) ice/snow for the (a,d,g) 800 m, (b,e,h) 1500 m, and (c,f,i) 3000 m cold temperature experiments with a CDNC of  $200\text{ cm}^{-3}$ .

**Figure 8.** Distribution of precipitation accumulated over the 12 hour simulations in (a) warm experiments and (b) cold experiments. Corresponding  $\delta^{18}\text{O}_{precip}$  for (c) warm experiments and (d) cold experiments. All profiles are based on the  $200\text{ cm}^{-3}$  simulations. The mountain peak is located at 300 km. Note that the axis limits for  $\delta^{18}\text{O}_{precip}$  have been shifted by 8 ‰ from the warm to cold simulations to account for the difference in the isotopic composition of vapor at the surface in the two cases.

Case	CDNC	Total Prec	Snow	Graupel	Rain	Spillover
W800m	25	38.6mm	7.8%	7%	85.2%	37%
W800m	100	32.1mm	13.5%	12.9%	73.6%	59%
W800m	200	28.5mm	14.3%	13.7%	72%	56%
W800m	400	23.7mm	14.8 %	14.2%	70.8%	62%
W800m	800	18.3mm	13.8%	13%	73.1%	68%
C800m	25	34.2mm	92.2%	2.6%	5.2%	45%
C800m	100	32.2mm	96.3%	3%	0.7%	50%
C800m	200	30.9mm	96.9%	2.9%	0.2%	53%
C800m	400	29.3mm	97.5%	2.5%	–	56%
C800m	800	26.6mm	98.1%	1.9%	–	61%
W1500m	25	113.6mm	60.9%	7%	32.1%	31%
W1500m	100	110.5mm	64.1%	11%	24.9%	33%
W1500m	200	106.6mm	65.8%	13%	21.2%	35%
W1500m	400	104.8mm	67%	13.7%	19.3%	36%
W1500m	800	105.3mm	67.1%	14%	18.9%	37%
C1500m	25	91.3mm	96.1%	0.5%	3.4%	32%
C1500m	100	92.6mm	98.3%	0.6%	1.1%	33%
C1500m	200	90.7mm	98.7%	0.7%	0.6%	34%
C1500m	400	91.7 mm	99.1%	0.7%	0.2%	35%
C1500m	800	87.3mm	99.4%	0.6%	–	38%
W3000m	25	235.8mm	67.8%	3.2%	29%	27%
W3000m	100	237.8mm	69.1%	3.8%	27.1%	27%
W3000m	200	233.9mm	71%	4.3%	24.7%	28%
W3000m	400	232.1mm	72.1%	4.8%	23.1%	29%
W3000m	800	227.9mm	74%	5.3%	20.7%	31%
C3000m	25	182.3mm	97.3%	0.5%	2.2%	20%
C3000m	100	183.2mm	98.7%	0.3%	1%	21%
C3000m	200	166.7mm	99%	0.3%	0.7%	22%
C3000m	400	175.1mm	99.4%	0.3%	0.3%	22%
C3000m	800	172.2mm	99.6%	0.3%	0.03%	22%

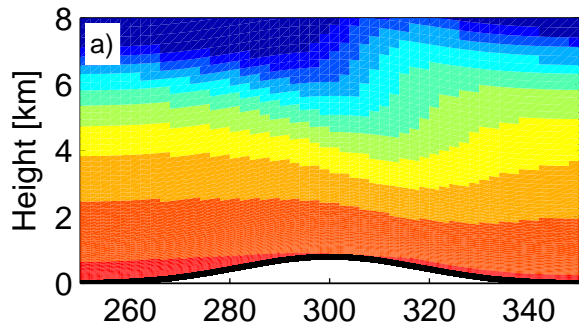
**Table 1:** Breakdown of the major statistics for all runs. Columns indicate the case (height and cold (C) or warm (W) initial temperature profile), cloud droplet number concentration (CDNC) in  $\text{cm}^{-3}$ , normalized (by the mountain half-width) domain-integrated precipitation over 12 hour simulation, and the percent of snow, graupel and rain. Spillover is the ratio of the total leeward precipitation to the total mountain precipitation.

**Figure 9.** Breakdown of (a) total precipitation, snow and graupel and (c) corresponding  $\delta^{18}\text{O}$  values for all experiments. The contributions from (b) precipitation sources normalized by the total precipitation and (d) the corresponding isotopic signatures for all experiments. Sources from left to right are: autoconversion of cloud liquid (LAUT), accretion of cloud liquid by rain (LACC), autoconversion of cloud ice (IAUT), accretion of cloud ice (IACC), riming of cloud liquid (RIM), vapor deposition onto ice (DEP), and sublimation of ice/evaporation of rain (SUB). All values are based on the  $200\text{ cm}^{-3}$  simulations.

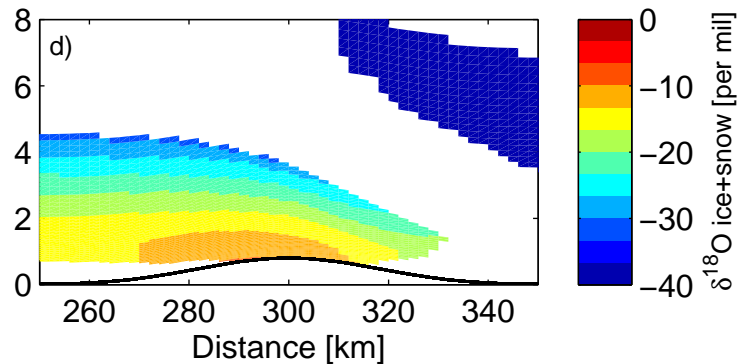
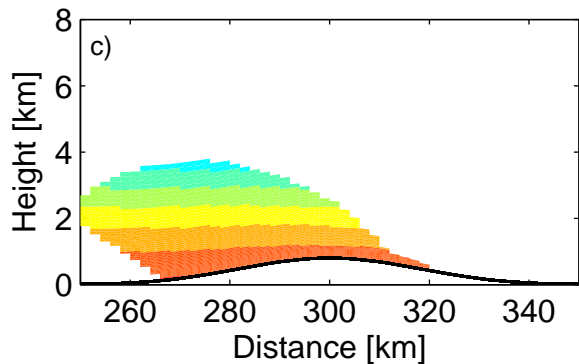
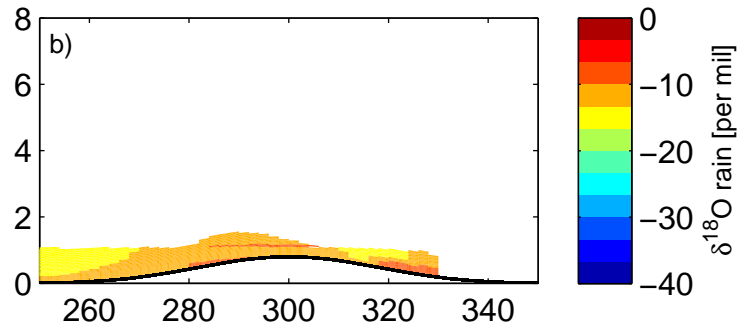
**Figure A1.** [Left panel:] the saturation ratios with respect to cloud liquid (blue) and ice (green) as a function of temperature. [Right panel:]  $\delta^{18}\text{O}$  of in-cloud water vapor (black), cloud liquid (blue) and vapor deposition onto ice particles (green) as a function of temperature.



W800m, CDNC=200 cm<sup>-3</sup>

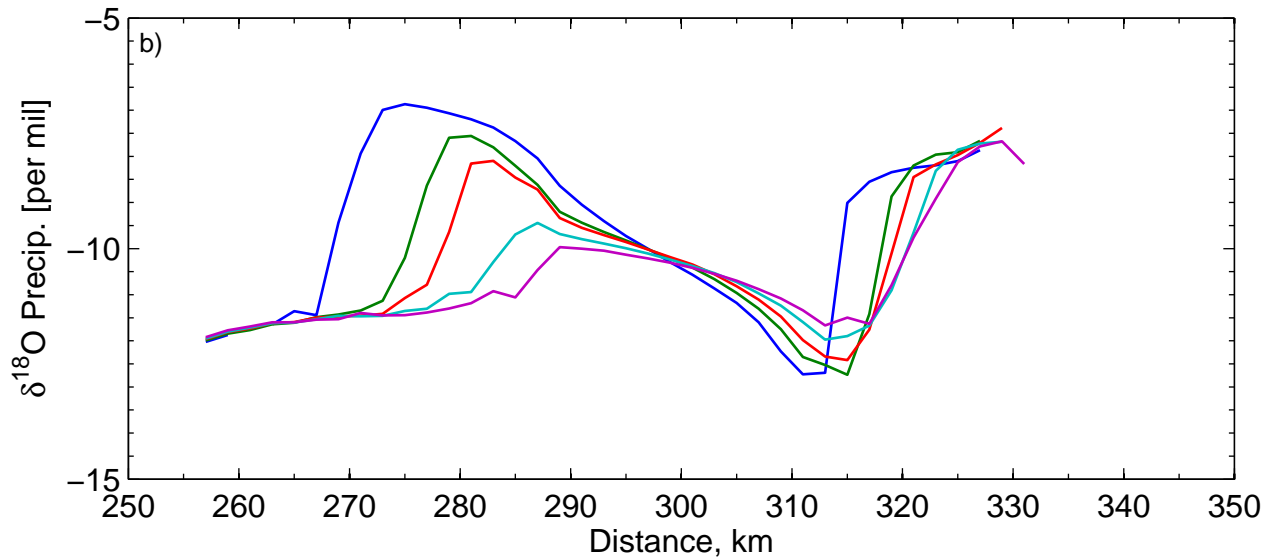
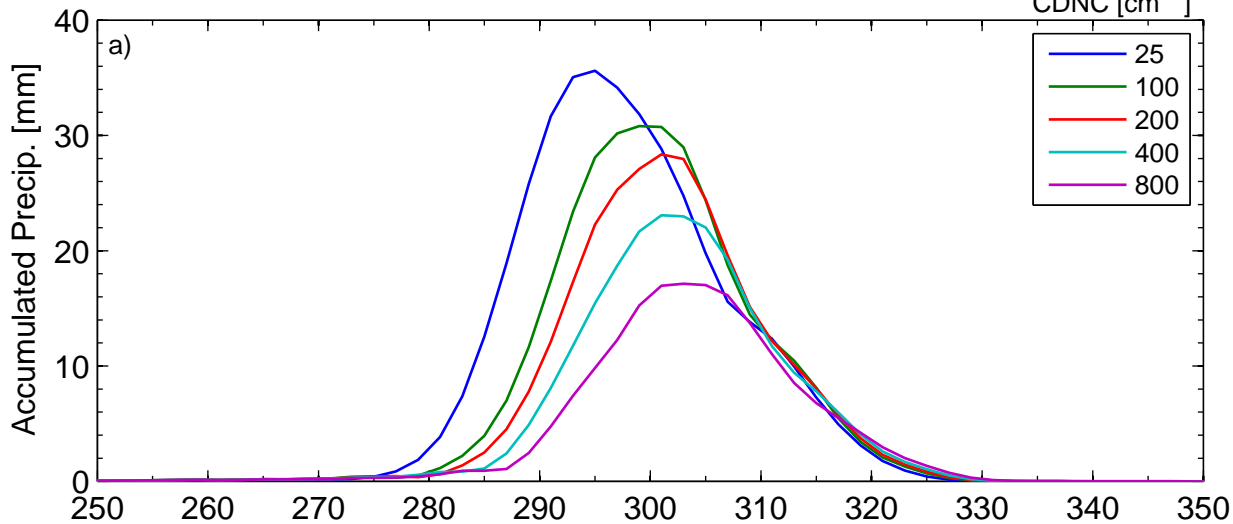


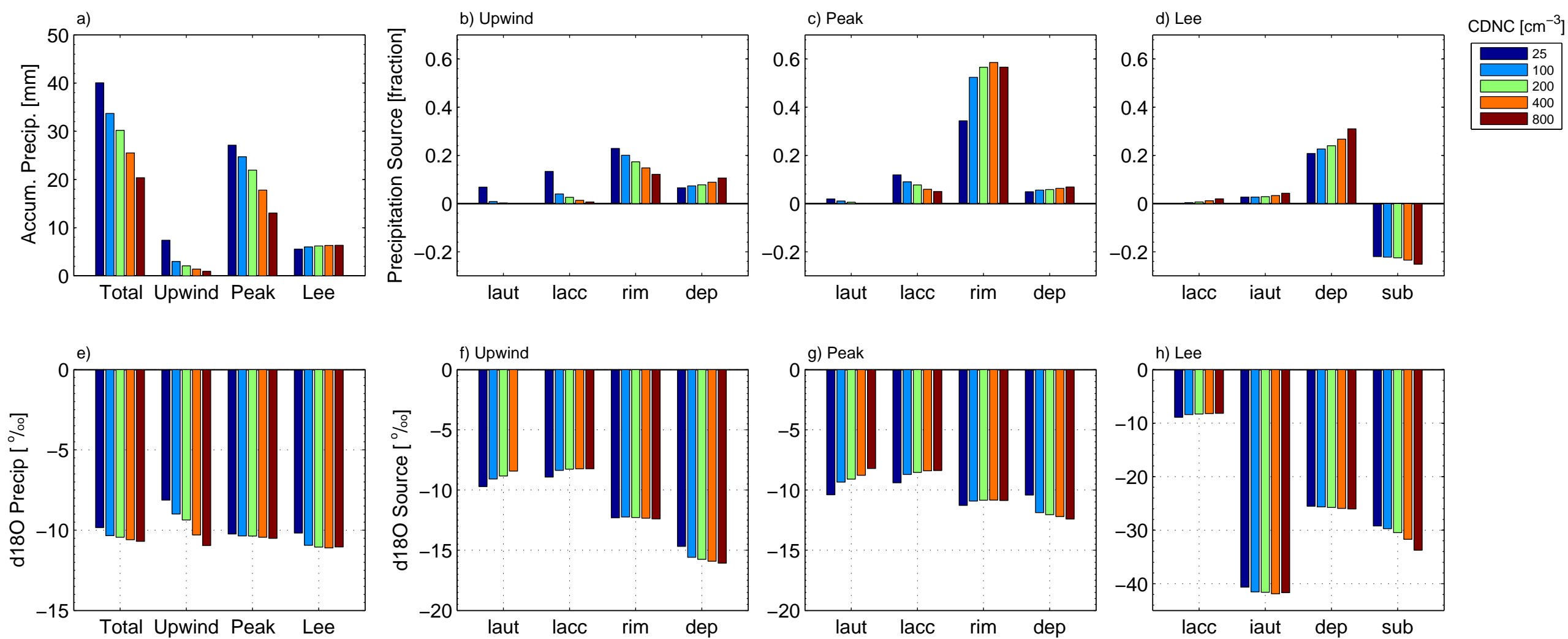
W800m, CDNC=200 cm<sup>-3</sup>

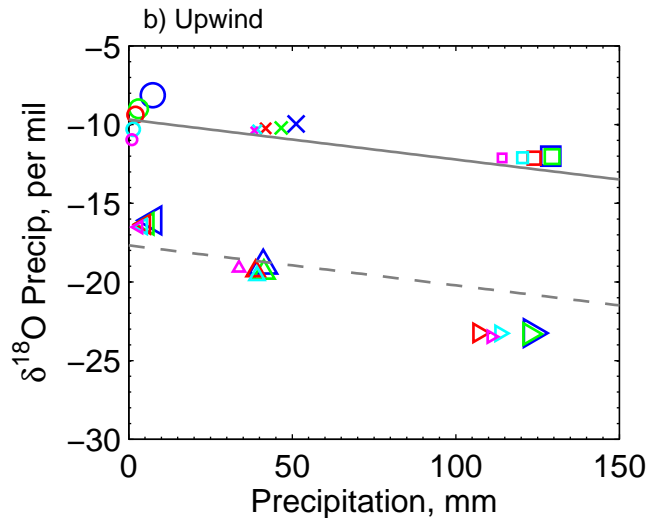
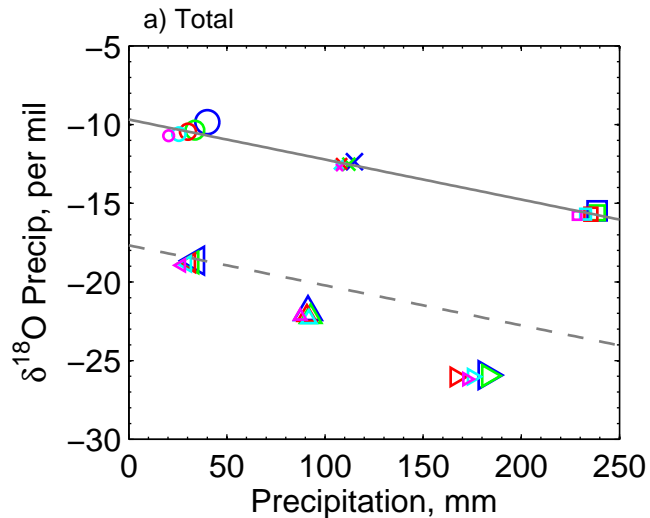




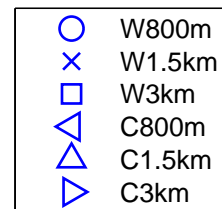
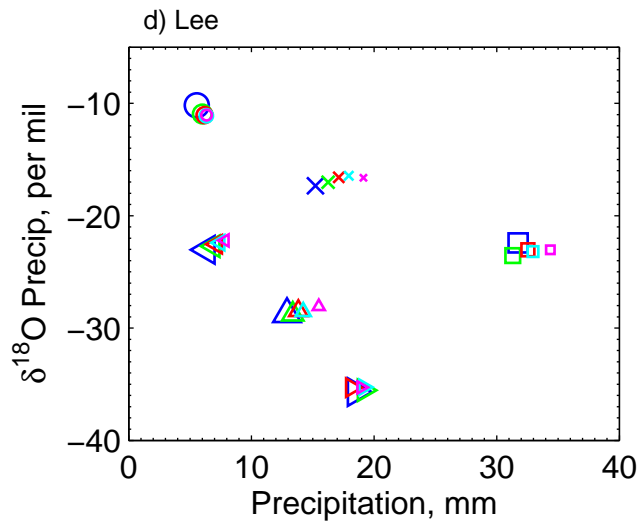
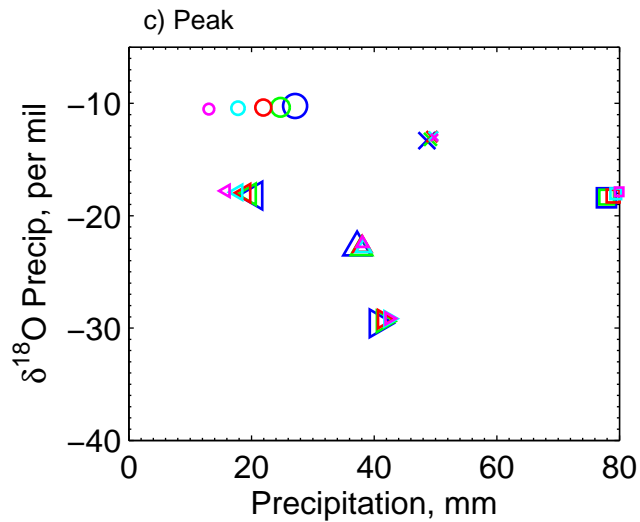
W800m

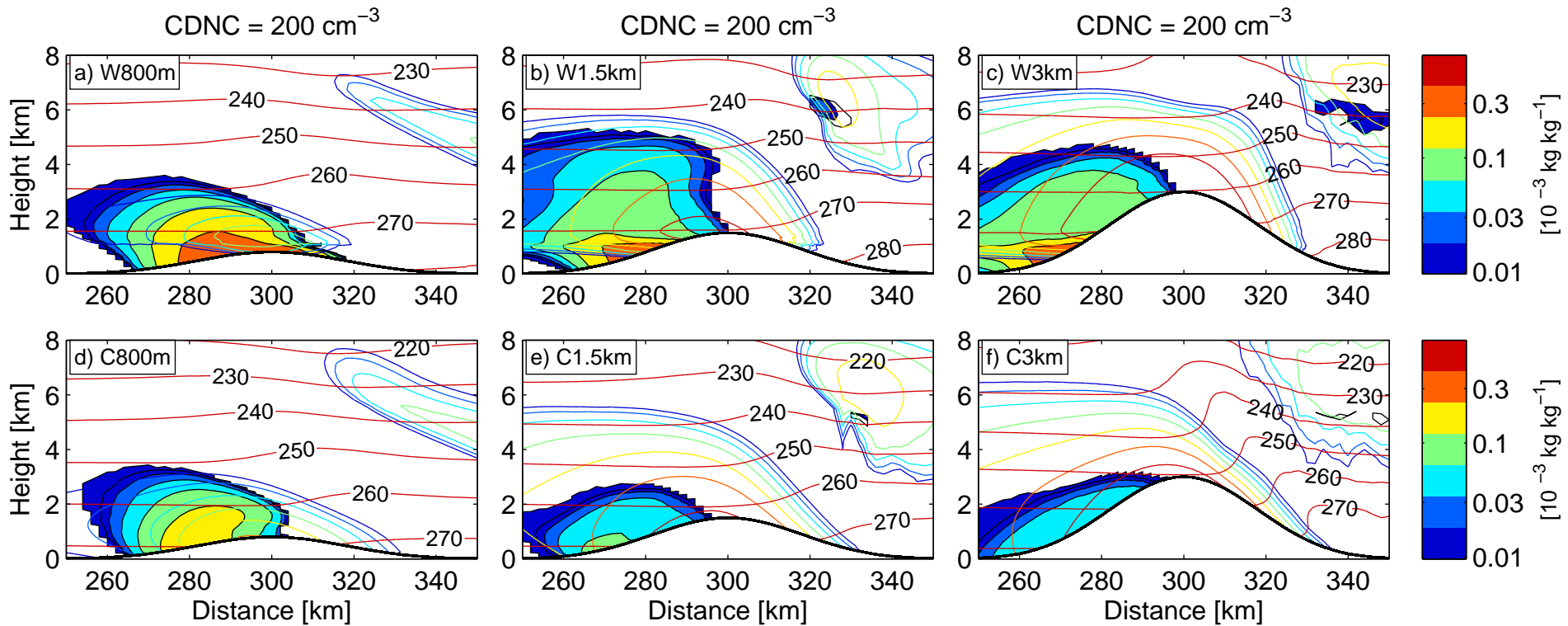
CDNC [cm<sup>-3</sup>]

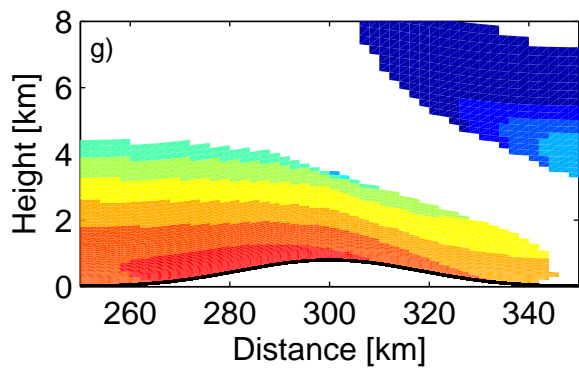
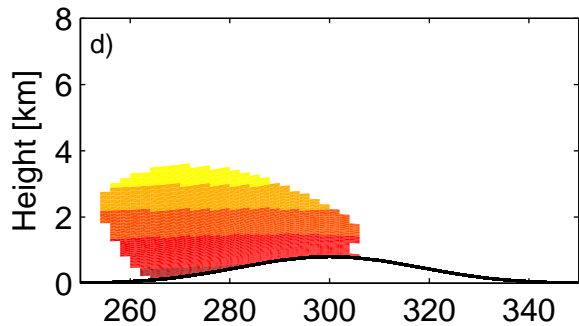
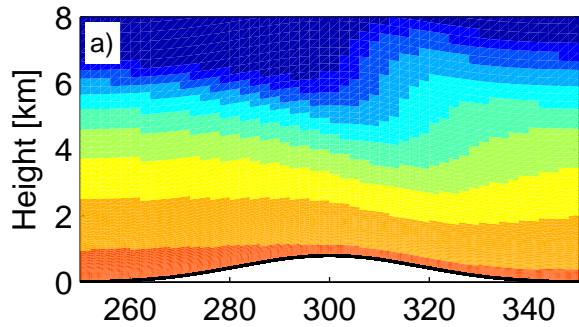
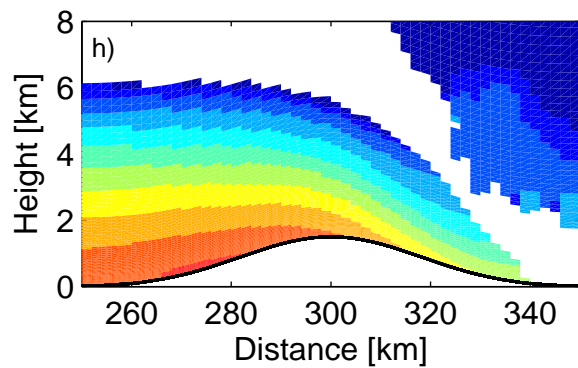
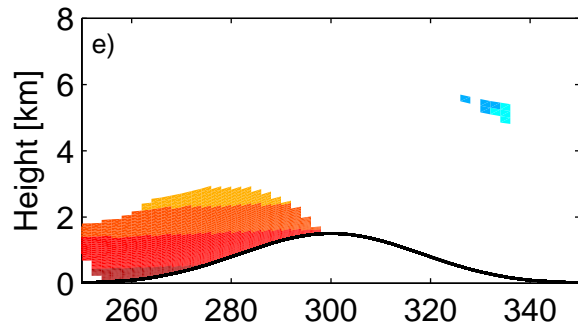
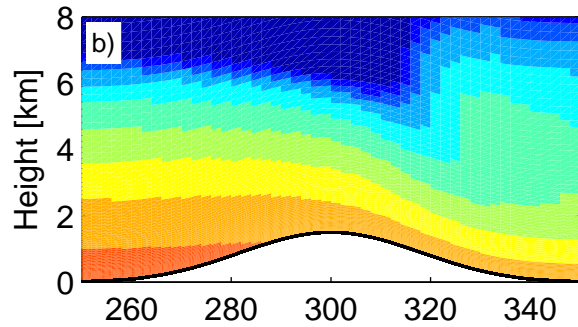
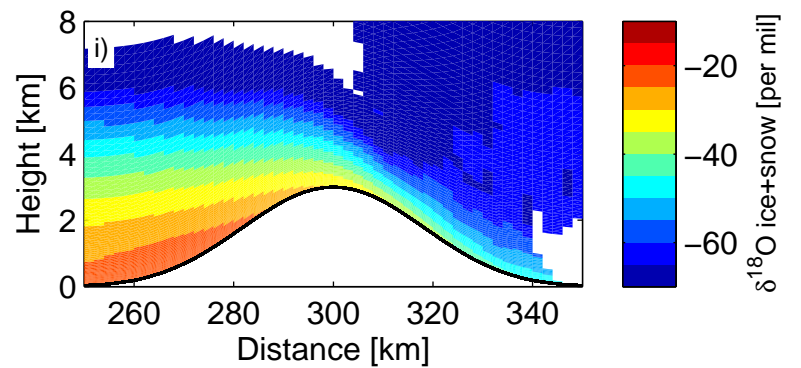
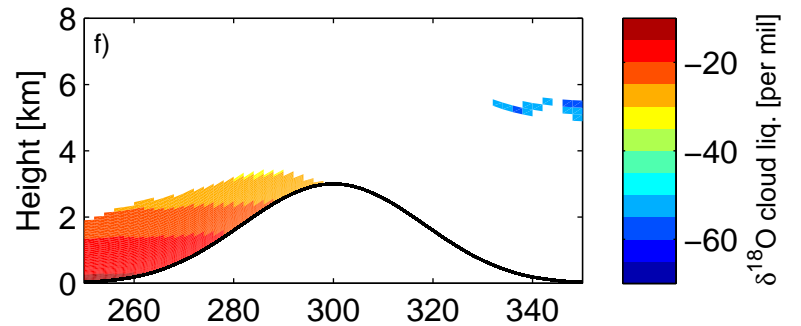
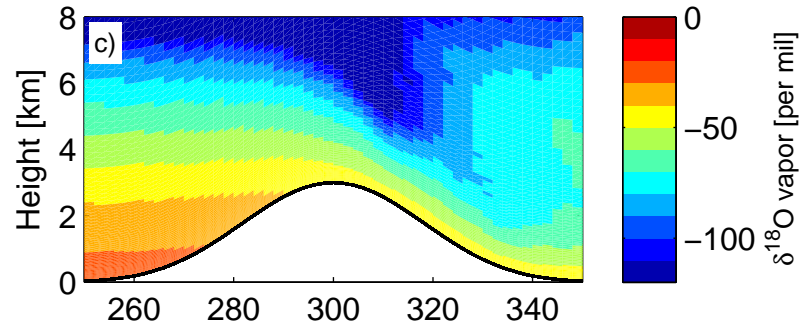


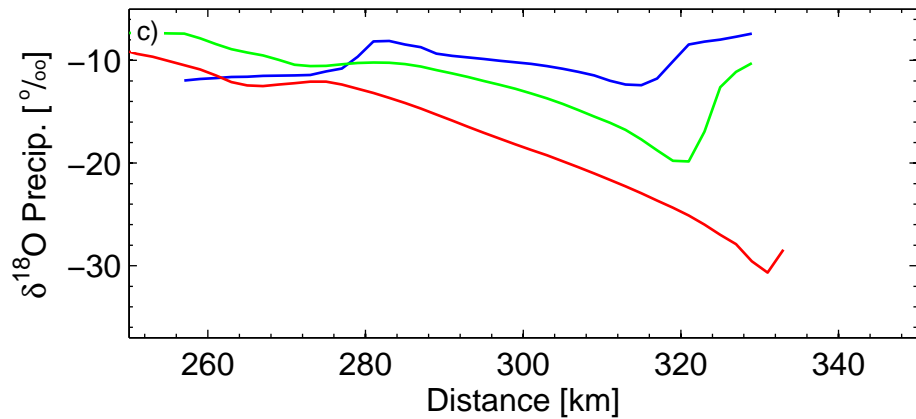
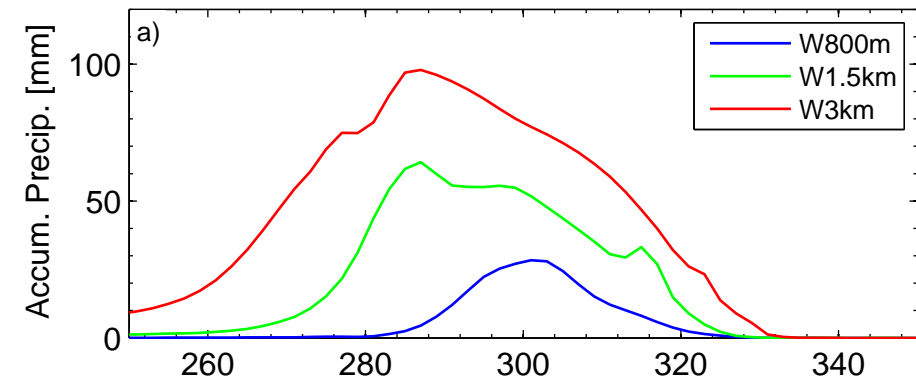
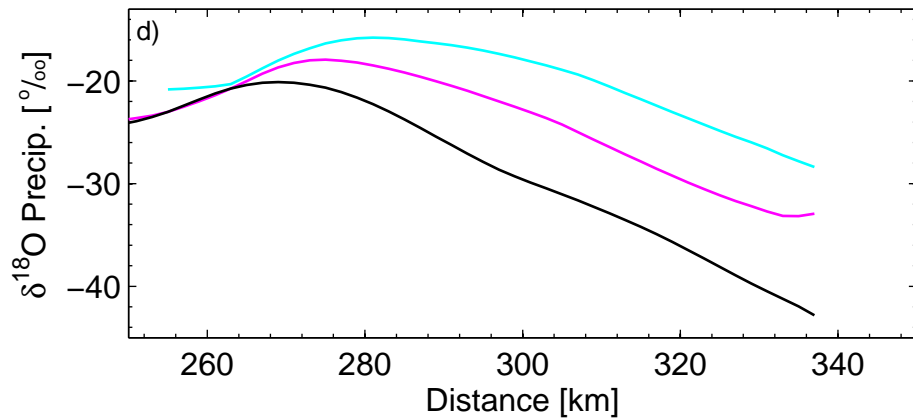
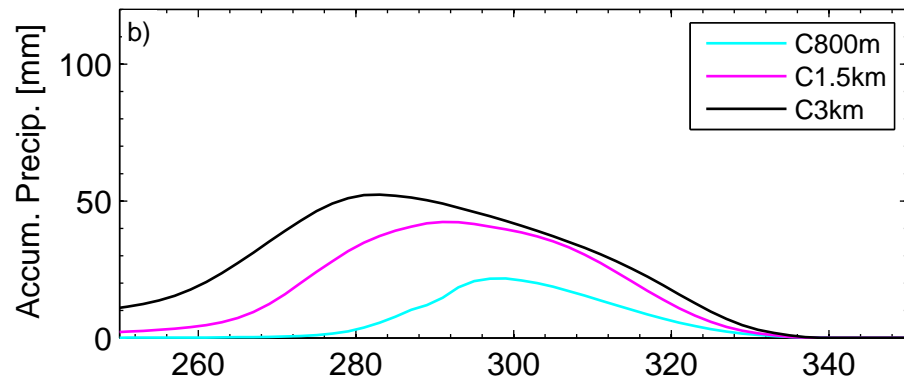


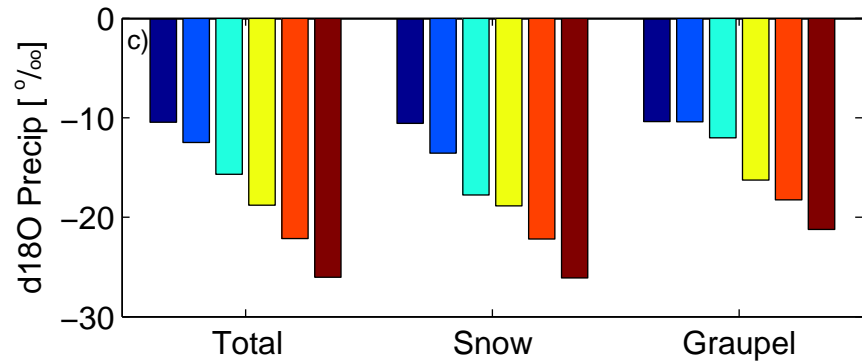
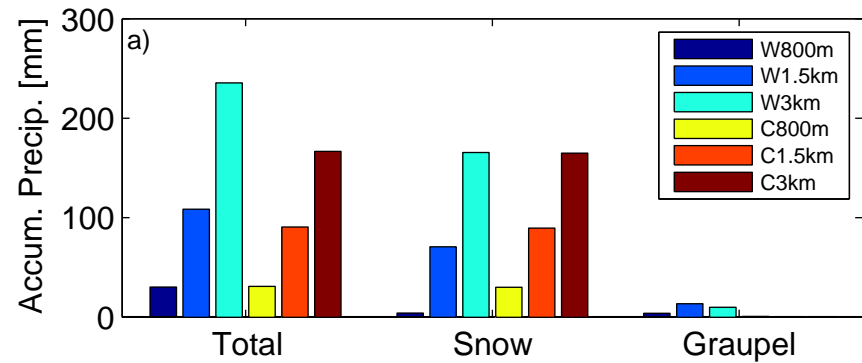
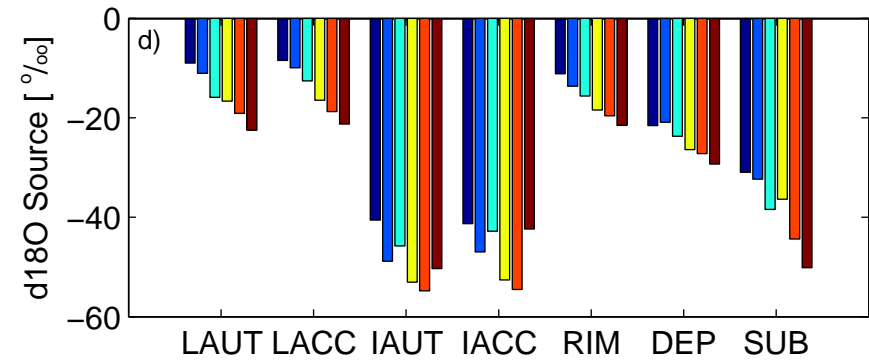
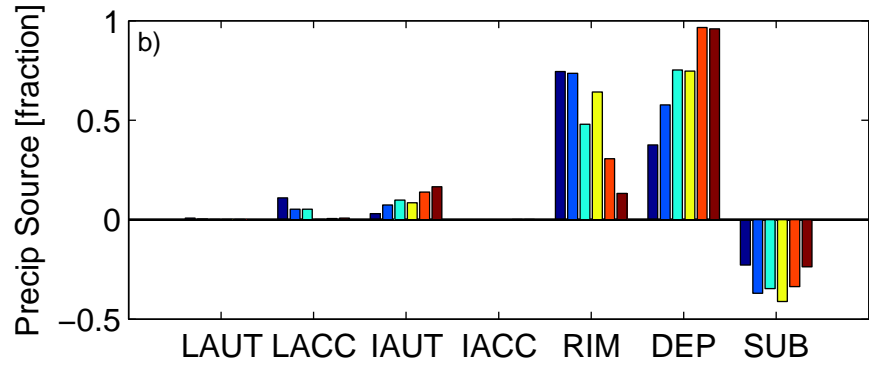
CDNC [ $\text{cm}^{-3}$ ]



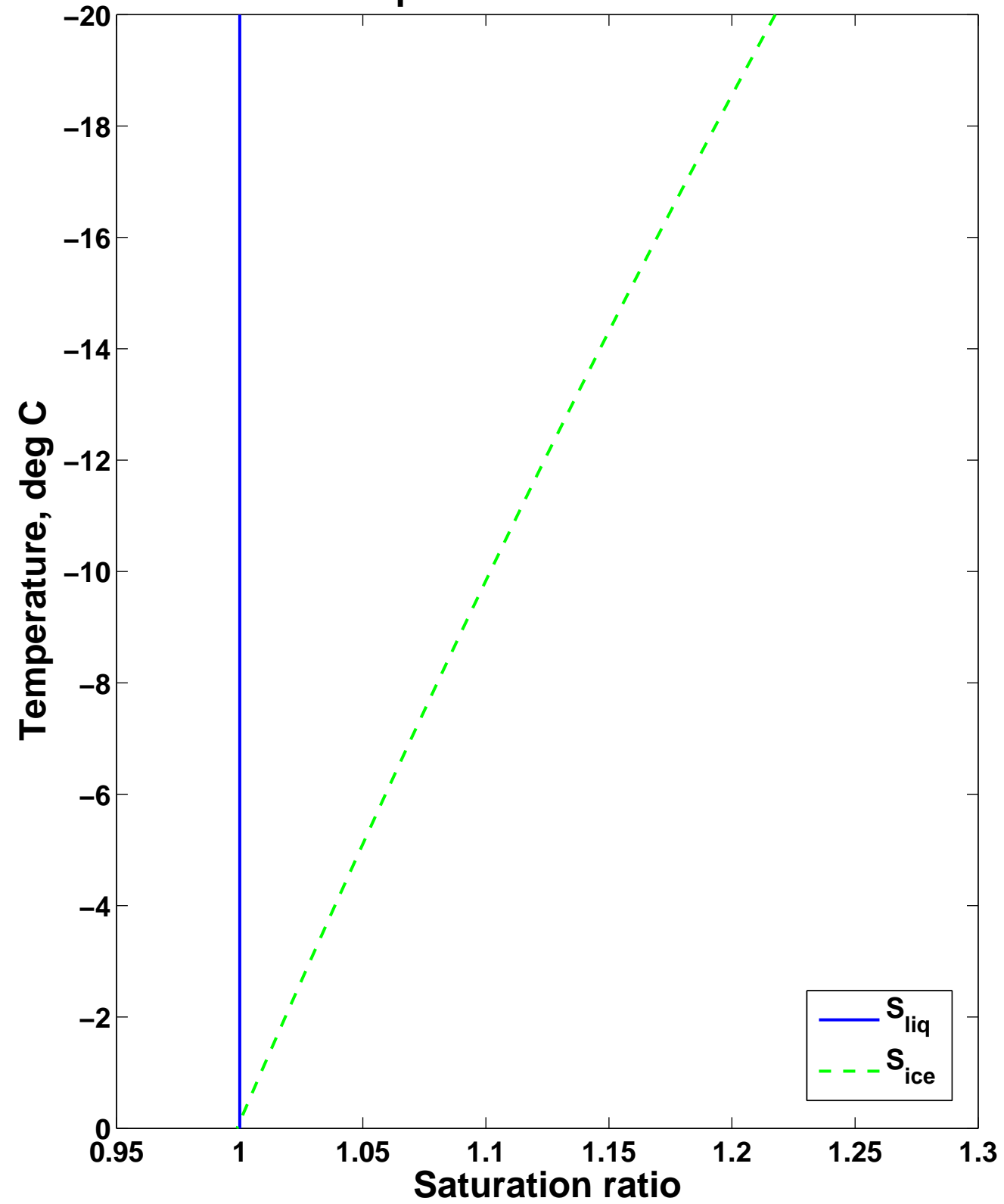


C800m, CDNC=200  $\text{cm}^{-3}$ C1.5km, CDNC=200  $\text{cm}^{-3}$ C3km, CDNC=200  $\text{cm}^{-3}$ 

CDNC=200 cm<sup>-3</sup>CDNC=200 cm<sup>-3</sup>

CDNC = 200 cm<sup>-3</sup>CDNC = 200 cm<sup>-3</sup>

### Cloud Liquid and Ice Saturation Ratios



### Ice, Vapor and Cloud Liquid $\delta^{18}\text{O}$

